Chemistry

Class-9

Chapter-6

Concept of Mole and Chemical counting

Subject teacher- Syeeda Sultana Work sheet -1 Date-09.08,2020

Exercise:

Write down the answers of the following questions on your copy.

Questions:

- 1. What is mole?
- **2.** What is Avogadro number?
- 3. Determine the molecular mass of Na₂SO₄?
- **4.** What is the relative molecular mass of glucose?
- **5.** What is the mass of $1 H_2O$ molecule?
- **6.** What is the mass of 5 water molecules?
- 7. How many H_2SO_4 molecules are there in $1g H_2SO_4$?
- **8.** Calculate the mole of H_2O present in $5g H_2O$.
- 9. Determine the number of H,S and O atoms in 1g H₂SO₄.
- 10. What is the total number of atoms present in 1 gram of methane (CH_4) ?
- 11. What is the mass of 3.01×10^{23} atoms of carbon?