## Chemistry Class-9 Chapter-6 Concept of Mole and Chemical counting Subject teacher- Syeeda Sultana Work sheet -6 Date-23.09.2020

## **Exercise:**

Write down the answers of the following questions on your copy.

## **Questions:**

- 1. What is chemical reaction and chemical equation?
- 2. What is balanced chemical equation?
- 3. What is reactants and products?
- 4. Why balancing of a chemical equation should be needed?
- 5. How can you write a complete chemical reaction between CaCO<sub>3</sub>& HCl?
- 6. Balancing & completing the following equations:
  - a)  $CaCO_{3(S)} + HClCaCl_{2(g)} + OO_{2(g)} + H_2O_{(l)}$
  - b) Mg (NO<sub>3</sub>)<sub>2 (S)</sub>  $\longrightarrow$  MgO (S) + ? + O<sub>2</sub>(g)
  - c)  $Al_2O_{3(s)} + HCl_{(aq)} \longrightarrow ?$
  - d) AgNO<sub>3 (s)</sub>  $\longrightarrow$  Ag<sub>(s)</sub> + NO<sub>2(g)</sub> + ?
  - e)  $Na_2CO_{3(S)} + HCl_{(aq)} \longrightarrow ?$
  - f) Na<sub>2</sub>CO<sub>3</sub> + 2CH<sub>3</sub>COOH  $\longrightarrow$  ?
  - g)  $2HCl + Na_2CO_3 \longrightarrow 2NaCl + ? + 2H_2O$
  - h)  $FeSO_4 + 2NaOH \longrightarrow ? + Na_2SO_4$